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1: HAPTICITY IN

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~~Important topics of~~
~~organic chemistry~~

~~Phenols-Organic~~
~~compounds containing~~

~~Oxygen-1~~ PART 3:

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METAL METAL BONDS IN ORGANOMETALLIC COMPOUNDS

Hybridization of Atomic
Orbitals, Sigma and Pi
Bonds, Sp Sp² Sp³,
Organic Chemistry,
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Rule of Law. Queen's
Law Faculty Research:
Prof. Karton ~~Don't study~~
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review) The 18 Electron Rule - Electron Counting via Method A and B + Hapticity Organolithium Reagents

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of 5: Grignard and organolithium reactions
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(CHM676) UiTMCS- 18
Electron Rule Metal
carbonyl back
bonding|IR stretching
frequency|Bonding in
metal olefin
complexes|OMC
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Organometallic Lec. 11
spectator ligand, back
bonding, donor &
acceptor ligand, CSIR,
GATE, IIT-JAM

Bridging ligands
Organometallic
compounds|Electron
contribution of bridging
ligands|Examples

Electron contribution of
ligands|Electron

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Organometallic
compounds|neutral ionic
methodChapter 13

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Chemistry. 13-7 Spectral
Analysis and

Characterization of

Organometallic

Complexes 13-4 Ligands
in Organometallic

Chemistry 13-5 Bonding

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Between Metal Atoms
and Organic Systems
13-6 Complexes
Containing M-C, M=C,
and M-C Bonds 13-3
The 18-Electron Rule
13-2 Organic Ligands
and Nomenclature 13-1
Historical Background.

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Chemistry. “ Inorganic
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Gary L. Miessler, Donald
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. Sandwich compounds

Cluster compounds. 13-1

Historical Background.

Other examples of
organometallic

compounds. 13-1

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Compound.

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chemistry is the study of
chemical compounds
containing bonds
between carbon and a
metal.

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aspects of inorganic

chemistry and organic

chemistry.

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compounds find

practical use in

stoichiometric and

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compounds. Electron counting is key in understanding organometallic chemistry. The 18-electron rule is helpful in predicting the stabilities of organometallic compounds. Organometallic compounds which have 18 electrons ...

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Section 2-13.3: Section

3-13.4: Section 4-13.5:

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e. Consider the complex. In the complex, atom has 8 electrons outside its noble gas core. Each is considered to act as a donor of 2 electrons,

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is considered to act 1
electron, each is
considered to act as a
donor of 2 electrons and
considered as a donor of
3 electrons. Thus, the
total electron count in
the complex is as follows:
Thus, is an 18-electron
complex.

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